nahMh

Rhodium Catalyzed Direct Arylation of α -Diazoimines

Dongari Yadagiri and Pazhamalai Anbarasan*

Department of Chemistry, Indian Institute of Technolo[gy](#page-3-0) Madras, Chennai − 600036, India

S Supporting Information

[ABSTRACT:](#page-3-0) An efficient rhodium catalyzed direct arylation of α -diazoimines, generated from readily accessible 1,2,3triazole, has been accomplished for the synthesis of 2,2-diaryl enamides. The reaction involves the chemo- and regioselective insertion of rhodium azavinyl carbene into aromatic $\mathrm{C}(sp^2)-\mathrm{H}$ bonds. Utility of the developed methodology was demonstrated in the synthesis of indole and tetrahydroisoquinoline frameworks.

T ransition metal catalyzed insertion of carbenes into various
C−H bonds is one of the most widely practiced methods in
organic symphosis¹. These resetions conservally show high organic synthesis.¹ These reactions generally show high selectivity for the intramolecular reaction with either a C(sp³)– H [o](#page-3-0)r $C(sp^2)$ -H bond, for the formation of small membered rings.² But, the intermolecular reaction of transition metal carbenes with C−H bonds is rather nonselective. Particularly, the reacti[o](#page-3-0)n of metal carbenes with arenes affords the cycloheptatriene, well-known as the Buchner reaction, 3 instead of the possible $C(sp^2)$ -H insertion (Scheme 1a).⁴

Scheme 1. Transition Metal Catalyzed React[io](#page-3-0)n of Diazo Compounds and Arenes

Recently, Wang et al.⁵ disclosed the Cu-catalyzed insertion of metal carbene generated from N-tosylhydrazones to the acidic C−H of 1,3-azoles. A [si](#page-3-0)milar reaction was later demonstrated employing either a Ni- or Co-based catalyst (Scheme 1b).⁶ Rhcatalyzed directing group assisted activation of the ortho C−H bond of arene followed by insertion of the diazo compoun[d](#page-3-0) was reported by Yu et al.⁷ and Li et al.⁸ (Scheme 1c). Subsequently,

the groups of Rovis, ⁹ Glorius,¹⁰ Cui,¹¹ and Wang¹² have demonstrated the study of diazo compounds in the Rh-catalyzed functionalization of var[io](#page-3-0)us direct[ing](#page-3-0) gro[up](#page-3-0) assisted C−[H b](#page-3-0)onds. These reactions are majorly limited to $C(sp^2)$ -H bonds of specific substrates, such as acidic heterocyclic or chelation assisted C−H bonds. Thus, the generally direct intermolecular insertion of transition metal carbenes into C−H bonds of simple arenes is highly warranted.

Use of 1,2,3-triazole as a source of α -diazoimines, which is difficult to access through traditional routes, 13 and its functionalization to various nitrogen-based building blocks and heterocycles, has gained reasonable attention in re[cen](#page-3-0)t years.¹⁴ Recently, we disclosed the Rh-catalyzed denitrogenative [2,3] sigmatropic rearrangement of azavinylcarbene derived fro[m](#page-3-0) 1,2,3-triazole with allyl aryl(alkyl) sulfides.¹⁵ Based on our interest in the C−H functionalization of arenes¹⁶ and unique reactivity of 1,2,3-triazole, we herein reveal th[e n](#page-3-0)ew Rh-catalyzed direct arylation of α-diazoimines generated fr[om](#page-3-0) N-sulfonyl-1,2,3-triazole with arenes (Scheme 1d).

At the beginning of our studies, we examined the rhodium acetate (2 mol %) catalyzed reaction of 1,2,3-triazoles (1 equiv) 1a with various arenes (4 equiv) in chloroform at 70 $^{\circ}$ C. Although a number of arenes such as xylene, mesitylene, and anisole did not show a promising result,¹⁷ N,N-diethylaniline 2a gave a detectable amount of product (21% yield) along with the hydrated product, α -aminoketone. A[na](#page-3-0)lysis of the isolated product proved the formation of enamide 4aa¹⁸ as a mixture of Z/E isomers in a 1.6:1 ratio, instead of expected imine 3, which may also arise from imine 3 through [im](#page-3-0)ine−enamine tautomerism (Scheme 2). The formation of product 4aa was further confirmed through hydrogenation over Pd/C to amide, where amide 5 was isol[at](#page-1-0)ed in 94% yield as the sole product.

The synthesis of a similar enamide was reported by Fokin and co-workers from 1,2,3-triazoles employing arylboronic acid, a prefunctionalized arene, as a coupling partner (Scheme 1d),¹⁹ but the present reaction utilizes the simple arene as a coupling

Received: March 24, 2014 Published: April 11, 2014

Scheme 2. Rh-Catalyzed Arylation of Triazole 1a with N,N-Diethylaniline 2a

partner and forms the enamide through C−H insertion. Interestingly, excellent selectivity is observed for the functionalization of para C−H bonds, which is not accessible through the chelation assisted strategy, over possible other C−H bonds in arenes and the known α -C(\overline{sp}^3)–H functionalization of alkylamines.²⁰

Encouraged by the result, various other critical parameters were investi[ga](#page-3-0)ted to improve the yield of 4aa. As shown in Table 1, increasing the temperature to 90 °C as well as changing the

Table 1. Rh-Catalyzed Arylation of Triazole 1a with 2a: Optimization^a

Ph	Ts 2a 1a	NEt ₂ Rh(II), solvent temp, 1 h	Ph NH 4aa Ťs	NEt ₂
entry	$Rh(II)$ -catalyst	solvent	temp $({}^{\circ}C)$	yield $(\%)^b$
1	$Rh_2(OAc)_4$	CHCl ₃	70	21 ^c
\mathfrak{p}	$Rh_2(OAc)_4$	DCE	90	52
3	$Rh_2(OAc)_4$	toluene	100	71 $(70)^d$
$\overline{4}$	$Rh_2(OAc)_4$	toluene	120	69
5	$Rh_2(OAc)_4$	C_6H_5Cl	120	46
6 ^e	$Rh_2(OAc)_4$	toluene	100	45
7	$Rh_2(Oct)_4$	toluene	100	70
8	$Rh_2(TBSP)_4$	toluene	100	34
9	$Rh_2(DOSP)_4$	toluene	100	49

a Triazole 1a (0.17 mmol, 1 equiv), arene 2a (0.66 mmol, 4 equiv), Rh(II)-catalyst (2 mol %) , solvent (1 mL) , temp, 1 h. b Isolated yields, Z/E ratio ∼1.6:1. ^c16 h. ^d1.7 mmol of 1a was used. ^e2 equiv of 2a.

solvent to 1,2-dichloroethane (DCE) showed a positive influence on the outcome of reaction and 4aa was isolated in 52% yield in 1 h (Table 1, entry 2). A similar trend was observed with toluene at 100 °C, where the formation of 4aa was observed in 71% yield (Table 1, entry 3). No arylated product derived from toluene was observed. However, a further change in either the temperature or solvent did not show any improvement (Table 1, entries 4 and 5). Screening other Rh-catalysts revealed $Rh_2(Oct)_4$ furnished 4aa in comparable yield (70%), but other bulky catalysts, $Rh_2(TBSP)_4$ and $Rh_2(DOSP)_4$, gave 4aa in 34% and 49% yield, respectively (Table 1, entries 7−9). Similarly, decreasing the equivalents of arene to 2 decreased the yield of 4aa (Table 1, entry 6).

Increasing the scale of the reaction by 10-fold furnished 4aa in 70% yield, which is highly important in the synthetic application (Table 1, entry 3). From the optimization studies, the following conditions were chosen for studying the generality of the present methodology: 1 equiv of 1,2,3-triazole 1a, 4 equiv of arene 2a, 2 mol % of $Rh_2(Oct)_4$, toluene, 100 °C, 1 h.

After identifying the optimized conditions, the generality of the present method was investigated with functionally different triazoles and arenes. As can be seen in Scheme 3, various

Scheme 3. Rh-Catalyzed Arylation of Triazoles 1 with 2a

substituted triazoles were subjected under the Rh-catalyzed arylation conditions with 2a to afford the enamides 4 as a mixture of isomers in a variable ratio (see Supporting Information). Changing the sulfonyl moiety of triazole to mesyl or benzenesulfonyl gave the corresponding enamide 4ba and 4ca in comparable yield. Alkyl (methyl, ethyl, tert[-butyl\)](#page-3-0) [substitute](#page-3-0)d aryl containing triazole underwent smooth arylation to furnish the enamides (4da−4ga) in good yield. Sterically hindered ortho substituted enamides (4ha and 4na) were synthesized in moderate to good yield. Interestingly, electron-rich anisyl substituted triazole also afforded the enamide 4ia in 70% yield. Synthetically useful halogen substituted aryl containing enamides (4ja−4na) were also synthesized in good yield. In addition, thiophene, a sulfur-containing heterocyclic substituted triazole, also tolerated the optimized conditions and afforded the enamide 4oa in 72% yield. Tetrasubstituted enamide 4pa was achieved in 75% yield from the corresponding 4,5-disubstituted triazole and 2a.

The Rh-catalyzed arylation of vinyl substituted triazole 1q and diethylaniline 2a under the optimized conditions furnished 6, as a single product (Scheme 4). The formation of 6 can be explained through the rearrangement of formed enamide to the conjugatively more stab[le](#page-2-0) azadiene.

Subsequently, the scope of arenes in the Rh-catalyzed arylation of triazole 1a was examined (Scheme 5). Symmetrically substituted aniline derivatives on reaction with 1a gave the

Scheme 4. Rh-Catalyzed Arylation of Triazoles 1q with 2a

Scheme 5. Rh-Catalyzed Arylation of Triazole 1a with Arenes 2

corresponding enamides 4ab, 4ac, and 4ad in 64%, 61%, and 51% yield, respectively. Unsymmetrically substituted nitrogen in aniline also afforded the enamides 4ae and 4af in good yield. Furthermore, meta-methyl and bromo substituted aniline derivatives furnished the corresponding enamides 4ag and 4ah in moderate to good yield. But, the ortho-substituted aniline derivatives did not afford the expected enamides (4ai and 4aj). Interestingly, the reaction of N-based heterocyclic arene (indole, indoline, tetrahydroquinoline, and dibenzoazepine), which contains the ortho-substitution in the fused form, with 1a giving the corresponding enamides (4ak−4an) in good yield. Furthermore, enamide 4ao was achieved from the arylation of 1a with a naphthalene derivative.

After establishing the scope of the developed methodology, the chemoselectivity of the Rh-catalyzed arylation reaction was investigated employing allyl substituted aniline derivatives, which is prone to other possible reactions such as cyclopropanation, [2,3]-sigmatropic rearrangement, and α -C−H insertion reactions. To our delight, the reaction of 1a under the Rh-catalyzed arylation conditions with 2p and 2q selectively afforded the $C(sp^2)$ –H inserted product, enamides 4ap and 4aq in excellent yield, respectively (Scheme 6). The structure of 4aq was unambiguously confirmed by X-ray analysis (see Supporting Information). 21 This demonstrates that the present method is highly selective for the arene C−H insertion over ot[her possible](#page-3-0) [reactions.](#page-3-0)

Scheme 6. Chemoselective Rh-Catalyzed Arylation of 1a

Enamides are present as key subunits in various natural products and pharmaceutically important molecules²² and are vital intermediates in organic synthesis having diverse synthetic utilities.²³ With the ready accessibility of various [e](#page-3-0)namides established, next, the synthetic application was demonstrated throug[h th](#page-3-0)e synthesis of a pharmaceutically important N-based heterocyclic system such as 3-arylindole and 4-arylisoquinolines (Scheme 7). 3-Arylindole 7 was achieved from the Cu-catalyzed

Scheme 7. Synthetic Utility of Enamide 4

C−N cross-coupling²⁴ of the ortho brominated enamide 4na. Consequently, hydrogenation of enamide 4aa over palladium on carbon afforded the [2](#page-3-0),2-diarylethylamide 5 in excellent yield. Protection of NH as a MOM group followed by TMSOTf mediated cyclization through an iminium ion furnished the 4 arylisoquinoline 8 in 47% overall yield.

A plausible mechanism for the direct arylation of triazole 1 with arene 2 to enamide 4 is shown in Scheme 8. The catalytic

Scheme 8. Plausible Mechanism

cycle starts with the formation of reactive rhodium carbenoid II from α -diazoimine I with the extrusion of nitrogen, which in turn is derived from triazole 1 via ring−chain isomerism. The formation of product 4 from II with an arene can be realized by two pathways. The concerted direct insertion of II into a C−H bond of arene would lead to the formation of 3, which upon tautomerization affords the enamide 4 (Path A). Alternatively,

the addition of 2 to an electrophilic metal carbene center with the assistance of a nitrogen lone pair would furnish the zwitterion III. The direct formation of 4 from III can be achieved by loss of a proton and rearomatization (Path B). Among them, path B can be a preferable pathway: (1) formation of III is favored with electron-rich arenes (aniline derivatives) over the neutral (toluene) and moderately electron-rich arene (anisole derivative), and (2) since involvement of a nitrogen lone pair is important to increase the nucleophilicity of the arene and formation of III, electrophilic substitution occurs at the para instead of the ortho position, and ortho-substituted (methyl and iodo) aniline derivatives were not successful under the present optimized conditions (Scheme 5). Similarly, attempts toward the functionalization of an ortho C−H bond with para-substituted N,N-diethyl-p-methylanilines [we](#page-2-0)re also unsuccessful. With the plausible mechanism, the present method complements and possesses an advantage over traditional acid mediated electrophilic aromatic alkylation/acylation, where electron-rich aniline derivatives are shown to be the least reactive.

In conclusion, we developed an efficient strategy for the direct arylation of azavinyl carbenes, derived from 1,2,3-triazole with various arenes. This strategy involves chemo- and regioselective arene $C(sp^2)$ – H insertion and offers electronically and sterically different substituted enamides, which are of high synthetic importance. Furthermore, the utility of enamides was demonstrated through the synthesis of N-based heterocycles, such as indole and isoquinolines.

■ ASSOCIATED CONTENT

S Supporting Information

Experimental methods, characterizations data, and $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectra of isolated compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: anbarasansp@iitm.ac.in.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

We thank Department of Science and Technology (DST), New Delhi for funding this work. D.Y. thanks IITM for an HTRA fellowship. We thank Mr. Ramkumar (IIT Madras) for single crystal analysis support.

■ REFERENCES

(1) For reviews on metal catalyzed carbene insertion into C−H bonds, see: (a) Davies, H. M. L.; Beckwith, R. E. Chem. Rev. 2003, 103, 2861. (b) Davies, H. M. L.; Manning, J. R. Nature 2008, 451, 417. (c) Doyle, M. P.; Duffy, R.; Ratnikov, M.; Zhou, L. Chem. Rev. 2010, 110, 704. (d) Davies, H. M. L.; Morton, D. Chem. Soc. Rev. 2011, 40, 1857.

(2) (a) Padwa, A.; Austin, D. J.; Price, A. T.; Semones, M. A.; Doyle, M. P.; Protopopova, M. N.; Winchester, W. R.; Tran, A. J. Am. Chem. Soc. 1993, 115, 8669. (b) Hashimoto, S.-I.; Watanabe, N.; Ikegami, S. J. Chem. Soc., Chem. Commun. 1992, 1508. (c) Tsutsui, H.; Yamaguchi, Y.; Kitagaki, S.; Nakamura, S.; Anada, M.; Hashimoto, S. Tetrahedron: Asymmetry 2003, 14, 817.

(3) Buchner, E.; Curtius, T. Ber. Dtsch. Chem. Ges. 1885, 18, 2377.

(4) For selected examples on Buchner reaction, see: (a) Maguire, A. R.; Buckley, N. R.; O'Leary, P.; Ferguson, G. Chem. Commun. 1996, 2595. (b) Park, C. P.; Nagle, A.; Yoon, C. H.; Chen, C.; Jung, K. W. J. Org. Chem. 2009, 74, 6231.

(5) Zhao, X.; Wu, G.; Zhang, Y.; Wang, J. J. Am. Chem. Soc. 2011, 133, 3296.

(6) Yao, T.; Hirano, K.; Satoh, T.; Miura, M. Angew. Chem., Int. Ed. 2011, 51, 775.

(7) Chan, W.-W.; Lo, S.-F.; Zhou, Z.; Yu, W.-Y. J. Am. Chem. Soc. 2012, 134, 13565.

(8) Yu, X.; Yu, S.; Xiao, J.; Wan, B.; Li, X. J. Org. Chem. 2013, 78, 5444. (9) Hyster, T. K.; Ruhl, K. E.; Rovis, T. J. Am. Chem. Soc. 2013, 135, 5364.

(10) Shi, Z.; Koester, D. C.; Boultadakis-Arapinis, M. l.; Glorius, F. J. Am. Chem. Soc. 2013, 135, 12204.

(11) Cui, S.; Zhang, Y.; Wang, D.; Wu, Q. Chem. Sci. 2013, 4, 3912.

(12) Hu, F.; Xia, Y.; Ye, F.; Liu, Z.; Ma, C.; Zhang, Y.; Wang, J. Angew. Chem., Int. Ed. 2014, 53, 1364.

(13) Gilchrist, T. L.; Gymer, G. E. Adv. Heterocycl. Chem. 1974, 16, 33. (14) For a highlight on 1,2,3-triazole as a source of azavinyl carbenes, see: (a) Gulevich, A. V.; Gevorgyan, V. Angew. Chem., Int. Ed. 2013, 52, 1371. For recent examples on functionalization of 1,2,3-triazole, see: (b) Chuprakov, S.; Worrell, B. T.; Selander, N.; Sit, R. K.; Fokin, V. V. J. Am. Chem. Soc. 2014, 136, 195. (c) Miura, T.; Tanaka, T.; Biyajima, T.; Yada, A.; Murakami, M. Angew. Chem., Int. Ed. 2013, 52, 3883. (d) Zibinsky, M.; Fokin, V. V. Angew. Chem., Int. Ed. 2013, 52, 1507. (e) Parr, B. T.; Davies, H. M. L. Angew. Chem., Int. Ed. 2013, 52, 10044. (f) Alford, J. S.; Spangler, J. E.; Davies, H. M. L. J. Am. Chem. Soc. 2013, 135, 11712. (g) Chuprakov, S.; Kwok, S. W.; Fokin, V. V. J. Am. Chem. Soc. 2013, 135, 4652. (h) Miura, T.; Tanaka, T.; Hiraga, K.; Stewart, S. G.; Murakami, M. J. Am. Chem. Soc. 2013, 135, 13652. (i) Parr, B. T.; Green, S. A.; Davies, H. M. L. J. Am. Chem. Soc. 2013, 135, 4716. (j) Schultz, E. E.; Sarpong, R. J. Am. Chem. Soc. 2013, 135, 4696. (k) Spangler, J. E.; Davies, H. M. L. J. Am. Chem. Soc. 2013, 135, 6802. (l) Miura, T.; Hiraga, K.; Biyajima, T.; Nakamuro, T.; Murakami, M. Org. Lett. 2013, 15, 3298. (m) Chuprakov, S.; Malik, J. A.; Zibinsky, M.; Fokin, V. V. J. Am. Chem. Soc. 2011, 133, 10352.

(15) Yadagiri, D.; Anbarasan, P. Chem.-Eur. J. 2013, 19, 15115.

(16) (a) Chaitanya, M.; Yadagiri, D.; Anbarasan, P. Org. Lett. 2013, 15, 4960. (b) Saravanan, P.; Anbarasan, P. Org. Lett. 2014, 16, 848.

(17) Recently, a reaction of α -diazoimines and arenes was reported to yield dearomatizing annulation in an intramolecular version, but no reaction was observed in an intermolecular version; see: Miura, T.; Funakoshi, Y.; Murakami, M. J. Am. Chem. Soc. 2014, 136, 2272.

(18) For an alternative synthesis of enamides, see: (a) $Goo\beta en, L. J.;$ Salih, K. S. M.; Blanchot, M. Angew. Chem., Int. Ed. 2008, 47, 8492. (b) Liwosz, T. W.; Chemler, S. R. Chem.-Eur. J. 2013, 19, 12771. (c) Gourdet, B.; Rudkin, M. E.; Watts, C. A.; Lam, H. W. J. Org. Chem. 2009, 74, 7849. (d) Vallin, K. S. A.; Zhang, Q.; Larhed, M.; Curran, D. P.; Hallberg, A. J. Org. Chem. 2003, 68, 6639. (e) Lee, J. M.; Ahn, D.-S.; Jung, D. Y.; Lee, J.; Do, Y.; Kim, S. K.; Chang, S. J. Am. Chem. Soc. 2006, 128, 12954. (f) Wallace, D. J.; Klauber, D. J.; Chen, C.-Y.; Volante, R. P. Org. Lett. 2003, 5, 4749.

(19) Selander, N.; Worrell, B. T.; Chuprakov, S.; Velaparthi, S.; Fokin, V. V. J. Am. Chem. Soc. 2012, 134, 14670.

(20) For selected examples on α -C−H insertion of ethers and amines, see: (a) Davies, H. M. L.; Beckwith, R. E. J.; Antoulinakis, E. G.; Jin, Q. J. Org. Chem. 2003, 68, 6126. (b) Davies, H. M. L.; Hansen, T.; Churchill, M. R. J. Am. Chem. Soc. 2000, 122, 3063. (c) Davies, H. M. L.; Grazini, M. n. V. A.; Aouad, E. Org. Lett. 2001, 3, 1475. (d) Davies, H. M. L.; Jin, Q. Org. Lett. 2004, 6, 1769.

(21) CCDC 979169 (4aq) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via http://www. ccdc.cam.ac.uk/data_request/cif.

(22) (a) Yet, L. Chem. Rev. 2003, 103, 4283. (b) Muller, K.; Sellmer, A.; Prinz, H. Eur. J. Med. Chem. 1997, 32, 895.

(23) (a) Matsubara, R.; Kobayashi, S. Acc. Chem. Res. 2008, 41, 292. (b) Carbery, D. R. Org. Biomol. Chem. 2008, 6, 3455.

(24) Surry, D. S.; Buchwald, S. L. Chem. Sci. 2010, 1, 13.